



US010675819B2

(12) **United States Patent**
Li et al.

(10) **Patent No.:** **US 10,675,819 B2**

(45) **Date of Patent:** ***Jun. 9, 2020**

(54) **MAGNETIC FIELD ALIGNMENT OF EMULSIONS TO PRODUCE POROUS ARTICLES**

(71) Applicants: **Massachusetts Institute of Technology**, Cambridge, MA (US); **Northeastern University**, Boston, MA (US)

(72) Inventors: **Linsen Li**, Malden, MA (US); **Jonathan Samuel Sander**, Zurich (CH); **Yet-Ming Chiang**, Weston, MA (US); **Randall Morgan Erb**, Newton, MA (US)

(73) Assignees: **Massachusetts Institute of Technology**, Cambridge, MA (US); **Northeastern University**, Boston, MA (US)

(*) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 687 days.

This patent is subject to a terminal disclaimer.

(21) Appl. No.: **15/390,295**

(22) Filed: **Dec. 23, 2016**

(65) **Prior Publication Data**

US 2017/0173893 A1 Jun. 22, 2017

Related U.S. Application Data

(63) Continuation-in-part of application No. 14/874,253, filed on Oct. 2, 2015.

(Continued)

(51) **Int. Cl.**

B29C 67/20 (2006.01)

B03C 1/32 (2006.01)

(Continued)

(52) **U.S. Cl.**

CPC **B29C 67/20** (2013.01); **B03C 1/32** (2013.01); **H01F 1/447** (2013.01); **H01M 4/04** (2013.01);

(Continued)

(58) **Field of Classification Search**

CPC **B29C 67/20**
See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

3,284,243 A 11/1966 Von Sturm

3,765,943 A 10/1973 Biagetti

(Continued)

FOREIGN PATENT DOCUMENTS

CA 2270771 10/2000

CA 2426156 C 4/2011

(Continued)

OTHER PUBLICATIONS

International Search Report and Written Opinion in International Patent Application No. PCT/US2015/053761, dated Feb. 18, 2016.

(Continued)

Primary Examiner — Monica A Huson

Assistant Examiner — Kelsey C Grace

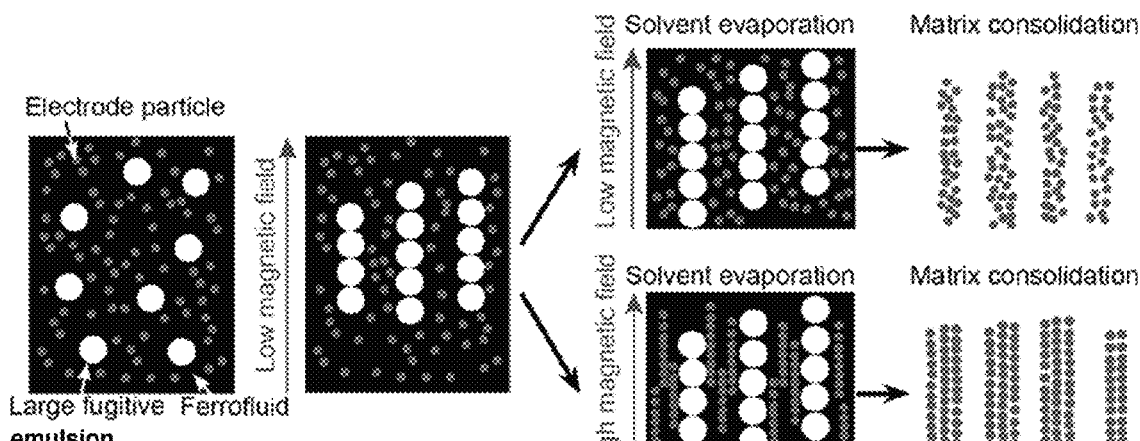
(74) *Attorney, Agent, or Firm* — Wolf, Greenfield & Sacks, P.C.

(57)

ABSTRACT

The use of magnetic fields in the production of porous articles is generally described. Certain embodiments are related to methods of producing porous articles in which magnetic fields are applied to an emulsion to align emulsion droplets. In some embodiments, after the emulsion droplets have been aligned, the emulsion droplets and/or the medium surrounding the emulsion droplets can be removed to leave behind a porous article. According to certain embodiments, polyvinyl alcohol can be used, for example, to stabilize the

(Continued)



emulsion droplets and/or bind together components of the porous article. In some embodiments, water-soluble liquid alcohol can be used, for example, to stabilize the suspension of electronically conductive material within a phase of the emulsion.

20 Claims, 18 Drawing Sheets

Related U.S. Application Data

- (60) Provisional application No. 62/347,475, filed on Jun. 8, 2016, provisional application No. 62/217,554, filed on Sep. 11, 2015, provisional application No. 62/059,315, filed on Oct. 3, 2014.

(51) Int. Cl.

H01M 4/525 (2010.01)
H01M 4/62 (2006.01)
H01M 4/04 (2006.01)
H01F 1/44 (2006.01)
H01M 4/131 (2010.01)
H01M 4/505 (2010.01)
H01M 4/1391 (2010.01)
H01M 4/587 (2010.01)
B29K 29/00 (2006.01)
B29K 105/00 (2006.01)
B29L 31/34 (2006.01)
H01M 4/02 (2006.01)

(52) U.S. Cl.

CPC *H01M 4/131* (2013.01); *H01M 4/1391* (2013.01); *H01M 4/505* (2013.01); *H01M 4/525* (2013.01); *H01M 4/587* (2013.01); *H01M 4/623* (2013.01); *H01M 4/625* (2013.01); *B29K 2029/04* (2013.01); *B29K 2105/0064* (2013.01); *B29K 2995/0005* (2013.01); *B29K 2995/0008* (2013.01); *B29L 2031/3468* (2013.01); *H01M 2004/021* (2013.01)

(56)

References Cited

U.S. PATENT DOCUMENTS

3,864,167 A 2/1975 Broadbead et al.
 4,245,016 A 1/1981 Rampel
 4,555,454 A 11/1985 Shuster et al.
 4,579,637 A 4/1986 Jaisinghani et al.
 4,599,114 A 7/1986 Atkinson
 4,615,784 A 10/1986 Stewart et al.
 4,668,596 A 5/1987 Shacklette et al.
 4,758,483 A 7/1988 Armand et al.
 4,804,592 A 2/1989 Vanderborgh et al.
 4,889,777 A 12/1989 Akuto
 5,100,747 A 3/1992 Hayashida et al.
 5,187,209 A 2/1993 Hirai et al.
 5,213,895 A 5/1993 Hirai et al.
 5,227,267 A 7/1993 Goebel et al.
 5,294,504 A 3/1994 Otagawa et al.
 5,338,625 A 8/1994 Bates et al.
 5,399,447 A 3/1995 Chaloner-Gill et al.
 5,405,937 A 4/1995 Lemaire et al.
 5,436,093 A 7/1995 Huang et al.
 5,441,830 A 8/1995 Moulton et al.
 5,464,707 A 11/1995 Moulton et al.
 5,478,676 A 12/1995 Turi et al.
 5,518,833 A 5/1996 Repplinger et al.
 5,520,850 A 5/1996 Chaloner-Gill et al.

5,567,754 A 10/1996 Stramel
 5,578,396 A 11/1996 Fauteux et al.
 5,587,253 A 12/1996 Gozdz et al.
 5,588,971 A 12/1996 Fauteux et al.
 5,589,297 A 12/1996 Koga et al.
 5,591,544 A 1/1997 Fauteux et al.
 5,624,605 A 4/1997 Cao et al.
 5,635,138 A 6/1997 Amatucci et al.
 5,645,781 A 7/1997 Popovic' et al.
 5,654,115 A 8/1997 Hasebe et al.
 5,677,080 A 10/1997 Chen
 5,698,342 A 12/1997 Klein
 5,705,291 A 1/1998 Amatucci et al.
 5,714,053 A 2/1998 Howard et al.
 5,733,683 A 3/1998 Searson et al.
 5,759,714 A 6/1998 Matsufuji et al.
 5,769,996 A 6/1998 McArdle et al.
 5,789,100 A 8/1998 Burroughs et al.
 5,821,033 A 10/1998 Cromack et al.
 5,827,615 A 10/1998 Touhsaent et al.
 5,834,136 A 11/1998 Gao et al.
 5,840,087 A 11/1998 Gozdz et al.
 5,843,292 A 12/1998 Spiros
 5,846,675 A 12/1998 Sazhin et al.
 5,888,670 A 3/1999 Kawakami
 5,897,522 A 4/1999 Nitzan
 5,902,689 A 5/1999 Vleggar et al.
 5,910,382 A 6/1999 Goodenough et al.
 5,925,283 A 7/1999 Taniuchi et al.
 5,949,213 A 9/1999 Lanni
 6,063,519 A 5/2000 Barker et al.
 6,063,525 A 5/2000 LaFollette
 6,069,221 A 5/2000 Chasser et al.
 RE36,843 E 8/2000 Lake et al.
 6,096,453 A 8/2000 Grunwald
 6,096,454 A 8/2000 Tran et al.
 6,096,494 A 8/2000 Tang et al.
 6,117,593 A 9/2000 Stachowiak et al.
 6,120,940 A 9/2000 Poehler et al.
 6,136,476 A 10/2000 Schutts et al.
 6,156,453 A 12/2000 Shimizu et al.
 6,159,389 A 12/2000 Miura et al.
 6,174,623 B1 1/2001 Shackle
 6,190,798 B1 2/2001 Okada et al.
 6,231,779 B1 5/2001 Chiang et al.
 6,242,124 B1 6/2001 Saito et al.
 6,280,875 B1 8/2001 Kwak et al.
 6,280,879 B1 8/2001 Andersen et al.
 6,287,720 B1 9/2001 Yamashita et al.
 6,291,097 B1 9/2001 Barker et al.
 6,300,016 B1 10/2001 Jan et al.
 6,306,540 B1 10/2001 Hiroi et al.
 6,322,924 B1 11/2001 Hirahara et al.
 6,337,156 B1 1/2002 Narang et al.
 6,342,317 B1 1/2002 Patel et al.
 6,358,645 B1 3/2002 Furukawa et al.
 6,376,393 B1 4/2002 Newton et al.
 6,395,429 B1 5/2002 Kang et al.
 6,403,263 B1 6/2002 Roach
 6,410,189 B1 6/2002 Yamada et al.
 6,413,284 B1 7/2002 Chu et al.
 6,413,285 B1 7/2002 Chu et al.
 6,432,585 B1 8/2002 Kawakami et al.
 6,451,487 B1 9/2002 Besner et al.
 6,454,977 B1 9/2002 Kwok et al.
 6,495,283 B1 12/2002 Yoon et al.
 6,511,780 B1 1/2003 Veregin et al.
 6,528,033 B1 3/2003 Barker et al.
 6,555,268 B1 4/2003 Inoue et al.
 6,565,763 B1 5/2003 Asakawa et al.
 6,599,662 B1 7/2003 Chiang et al.
 6,645,675 B1 11/2003 Munshi
 6,664,006 B1 12/2003 Munshi
 6,730,438 B2 5/2004 Nakanishi et al.
 6,753,111 B2 6/2004 Kweon et al.
 6,764,525 B1 7/2004 Whitacre et al.
 6,783,894 B2 8/2004 Kajjura et al.

(56)

References Cited

U.S. PATENT DOCUMENTS

6,800,399 B2 10/2004 Matsumoto
 6,818,356 B1 11/2004 Bates
 6,844,105 B1 1/2005 Hanafusa et al.
 6,855,273 B2 2/2005 Ravet et al.
 6,878,487 B2 4/2005 Cho et al.
 6,982,132 B1 1/2006 Goldner et al.
 7,026,071 B2 4/2006 Mayes et al.
 7,033,702 B2 4/2006 Dasgupta et al.
 7,087,348 B2 8/2006 Holman et al.
 7,090,785 B2 8/2006 Chiang et al.
 7,318,982 B2 1/2008 Gozdz et al.
 7,338,734 B2 3/2008 Chiang et al.
 7,387,851 B2 6/2008 Gozdz et al.
 7,553,584 B2 6/2009 Chiang et al.
 7,579,112 B2 8/2009 Chiang et al.
 7,662,265 B2 2/2010 Chiang et al.
 7,763,382 B2 7/2010 Riley et al.
 7,781,098 B2 8/2010 Chiang et al.
 7,988,746 B2 8/2011 Chiang et al.
 8,148,009 B2 4/2012 Chiang et al.
 8,168,326 B2 5/2012 Chiang et al.
 8,277,975 B2 10/2012 Chiang et al.
 9,065,093 B2 6/2015 Chiang et al.
 10,164,242 B2 12/2018 Chiang et al.
 2001/0005558 A1 6/2001 Yoshioka et al.
 2001/0005562 A1 6/2001 Yoshioka et al.
 2001/0007726 A1 7/2001 Yoshioka et al.
 2001/0043709 A1 11/2001 Panitzsch
 2002/0000034 A1 1/2002 Jensen
 2002/0015278 A1 2/2002 Fukuyama et al.
 2002/0028380 A1 3/2002 Tanjo et al.
 2002/0036282 A1 3/2002 Chiang et al.
 2002/0048706 A1 4/2002 Mayes et al.
 2002/0071990 A1 6/2002 Kweon et al.
 2002/0071991 A1 6/2002 Kweon et al.
 2002/0074972 A1 6/2002 Narang et al.
 2002/0192137 A1 12/2002 Chaloner-Gill et al.
 2002/0195591 A1 12/2002 Ravet et al.
 2003/0003352 A1 1/2003 Kweon et al.
 2003/0008212 A1 1/2003 Akashi et al.
 2003/0049529 A1 3/2003 Cho et al.
 2003/0054250 A1 3/2003 Kweon et al.
 2003/0082446 A1 5/2003 Chiang et al.
 2003/0099884 A1 5/2003 Chiang et al.
 2003/0114297 A1 6/2003 Shinn et al.
 2004/0005265 A1 1/2004 Chiang et al.
 2004/0018429 A1 1/2004 Kweon et al.
 2004/0018430 A1 1/2004 Holman et al.
 2004/0018431 A1 1/2004 Gozdz et al.
 2004/0151887 A1 8/2004 Forrest et al.
 2004/0185343 A1 9/2004 Wang et al.
 2004/0265692 A1 12/2004 Long et al.
 2005/0026037 A1 2/2005 Riley, Jr. et al.
 2005/0034993 A1 2/2005 Gozdz et al.
 2005/0048374 A1 3/2005 Yonezawa
 2005/0082726 A1 4/2005 Hilmas et al.
 2005/0175529 A1 8/2005 Ceder et al.
 2005/0181280 A1 8/2005 Ceder et al.
 2005/0272214 A1 12/2005 Chiang et al.
 2006/0035143 A1 2/2006 Kida et al.
 2006/0102455 A1 5/2006 Chiang et al.
 2006/0234131 A1 10/2006 Takami et al.
 2007/0009801 A1 1/2007 Inagaki et al.
 2007/0275300 A1 11/2007 Salot et al.
 2008/0099734 A1 5/2008 Chiang et al.
 2008/0213662 A1 9/2008 Chiang et al.
 2008/0254368 A1 10/2008 Ooyama et al.
 2008/0292862 A1 11/2008 Filippov et al.
 2008/0311470 A1 12/2008 Gozdz et al.
 2009/0035664 A1 2/2009 Chiang et al.
 2009/0202903 A1 8/2009 Chiang et al.
 2009/0235520 A1 9/2009 Atsumi et al.
 2009/0253025 A1 10/2009 Whitacre

2010/0248028 A1 9/2010 Riley et al.
 2011/0005065 A1 1/2011 Chiang et al.
 2011/0038656 A1 2/2011 Chiang et al.
 2011/0045346 A1 2/2011 Chiang et al.
 2011/0070489 A1 3/2011 Chiang et al.
 2011/0097623 A1 4/2011 Chiang et al.
 2011/0151324 A1 6/2011 Chiang et al.
 2011/0278170 A1 11/2011 Chiang et al.
 2012/0315538 A1 12/2012 Chiang et al.
 2015/0364753 A1 12/2015 Chiang et al.
 2016/0096334 A1* 4/2016 Sander B29C 67/202
 264/427
 2017/0087074 A1* 3/2017 Perusse A61K 8/416

FOREIGN PATENT DOCUMENTS

CN 1470083 A 1/2004
 CN 1901255 A 1/2007
 CN 1945881 A 4/2007
 DE 102007042217 A 3/2009
 EP 0 757 407 A1 2/1997
 EP 967672 A1 12/1999
 EP 1 113 511 A1 7/2001
 EP 1 042 832 A1 10/2001
 EP 1 184 918 3/2002
 EP 1 231 651 8/2002
 EP 1 231 653 8/2002
 EP 1 860 722 A1 8/2002
 GB 976971 12/1964
 JP 53-98038 8/1978
 JP 55-010738 A 1/1980
 JP 59-094379 5/1984
 JP 59-152565 A 8/1984
 JP 60-183365 A 9/1985
 JP 61-4167 A 1/1986
 JP 63-121272 A 5/1988
 JP 63-289768 11/1988
 JP 04-58455 2/1992
 JP 05-029006 2/1993
 JP 05-041211 A 2/1993
 JP 05-109429 4/1993
 JP 60-104167 A 4/1994
 JP 06-236768 A 8/1994
 JP 71-01728 A 4/1995
 JP 07-153492 A 6/1995
 JP 07-262986 A 10/1995
 JP 08-138650 A 5/1996
 JP 08-321306 A 12/1996
 JP 09-022693 A 1/1997
 JP 09-147862 6/1997
 JP 09-298057 A 11/1997
 JP 09-320569 A 12/1997
 JP 11-121006 4/1999
 JP 11-121025 A 4/1999
 JP 11-121061 4/1999
 JP 11-154508 6/1999
 JP 11-162519 A 6/1999
 JP 2000-021415 A 1/2000
 JP 2000-045639 A 2/2000
 JP 2000-080406 A 3/2000
 JP 2000-106213 4/2000
 JP 2000-173645 A 6/2000
 JP 2000-195525 A 7/2000
 JP 2000-195526 A 7/2000
 JP 2000-228213 A 8/2000
 JP 2000-248095 A 9/2000
 JP 2001-151834 6/2001
 JP 2002-151055 A 5/2002
 JP 4-206366 B2 1/2009
 KR 1019910007466 9/1991
 KR 1998-071228 10/1998
 WO WO 97/05666 A2 2/1997
 WO WO 97/27635 A1 7/1997
 WO WO 97/44843 11/1997
 WO WO 98/12761 A1 3/1998
 WO WO 98/16960 A2 4/1998
 WO WO 98/48466 A1 10/1998

(56)

References Cited

FOREIGN PATENT DOCUMENTS

WO	WO 00/10736 A1	3/2000
WO	WO 00/41256 A1	7/2000
WO	WO 01/77501 A2	10/2001
WO	WO 02/43168 A2	5/2002
WO	WO 03/012908 A2	2/2003
WO	WO 03/056646 A1	7/2003
WO	WO 07/028972 A1	3/2007
WO	WO 08/011061 A1	1/2008
WO	WO 08/153749	12/2008
WO	WO 2013/112670 A1	8/2013
WO	WO 14/170024 A1	10/2014
WO	WO 2016/054530 A1	4/2016

OTHER PUBLICATIONS

Aleshin, A.N. et al., "Transport Properties of Poly (3,4-Ethylenedioxythiophene)/Poly (Styrenesulfonate)," *Synthetic Metals* 94:173-7 (1998).

Allen, J.L., et al., "Li_{1-x}Fe_{1-x}PO₄: Electronically conductive Lithium Iron Phospho-olivines with Improved Electrochemical Performance," *The Proceedings of The Electrochemical Society* 28:198-205 (2003).

Arbizzani et al., "Li_{1.01}Mn_{1.974} surface modification by pol(3,4-ethylenedioxythiophene)," *J. Power Sources*, 2003;119-121:695-700.

Arbizzani et al., "Li_{1.03}Mn_{1.97}O₄ Surface Modification by Poly (3,4-Ethylenedioxythiophene)," Poster presented at 11th Intl. Meeting on Lithium Batteries, Jun. 23-28, 2002, Monterey, CA.

Armand et al., Conjugated dicarboxylate anodes for Li-ion batteries. *Nature Materials* Jan. 18, 2009. 8: 120-5. DOI: 10.1038/nmat2372.

Armand, M.B. et al., "Poly-Ethers as Solid Electrolytes," *Proceedings of the Intl. Conference on Fast Ion Transport in Solids, Electrodes and Electrolytes*, Lake Geneva, WI, 131-136 (1979).

Bae et al., Design of Battery Electrodes with Dual-Scale Porosity to Minimize Tortuosity and Maximize Performance. *Advanced Materials*. 2012. 5 pages. DOI: 10.1002/adma.201204055.

Bae et al., High density sintered electrode architecture to improve the energy density of rechargeable lithium batteries. Poster presented Jul. 1, 2010 at the 15th International Meeting on Lithium Batteries, Jun. 27-Jul. 2, 2010. Montreal, Quebec, Canada. 1 page.

Bae et al., Influence of controlled pore topology in sintered Li-ion battery cathodes on electrochemical performance. Presented at the Materials Research Society fall meeting in Nov. 2010. 18 pages.

Bervas et al., Investigation of the Lithiation and Delithiation Conversion Mechanisms in a Bismuth Fluoride Nanocomposites. *J. Electrochem. Soc.* Mar. 2, 2006. 153(4): A799-808.

Bouridah, A. et al., "Poly(Dimethylsiloxane)-Poly(Ethylene Oxide) Based Polyurethane Networks Used as Electrolytes in Lithium Electrochemical Solid State Batteries," *Solid State Ionics*, 15 (1985) pp. 233-240.

Broadhead et al., Chapter 2. *Electrochemical Principles and Reactions. Handbook of Batteries*. 1995:2.1-2.35.

Cernak et al., Aggregation dynamics of nonmagnetic particles in a ferrofluid. *Physical Review E*. Sep. 29, 2004;70(3):031504.

Chan et al., High-performance lithium battery anodes using silicon nanowires. *Nature Nanotechnology*. 2008. 3: 31-5. Doi:10.1038/nnano.2007.411. Published online Dec. 16, 2007.

Chen et al., Reducing Carbon in LiFePO₄/C Composite Electrodes to Maximize Specific Energy, Volumetric Energy, and Tap Density. *J Electrochem Soc.* 2002;148(9):A1184-89.

Chen-Wiegart et al., "Tortuosity characterization of 3D microstructure at nano-scale for energy storage and conversion materials," *Journal of Power Sources*. 2014;249:349-56.

Chiang et al., Electrode architectures for enhancing energy density in rechargeable lithium batteries. Spring MRS Meeting Invited Talk Abstract. Abstract available prior to Apr. 7, 2010. San Francisco, California. 1 page.

Chiang, Aligned (Unit Tortuosity) Porosity: 3 approaches using a removable (fugitive) pore former. Presentation slideshow accompanying talk given at Spring MRS Meeting Invited Talk. Apr. 7, 2010. No handouts were distributed. 11 pages.

Chiang, Y.-M. et al., "High Capacity, Temperature-Stable Lithium Aluminum Manganese Oxide Cathodes for Rechargeable Batteries," *Electrochem Sol St Lett*, 2 (3) (1999) pp. 107-110.

Chiang, Design and Scalable Assembly of High Density Low Tortuosity Electrodes. Massachusetts Institute of Technology, Cambridge, MA. Jun. 9, 2016. Believed to have been distributed on May 9, 2016. Presented on Jun. 9, 2016.

Chiang, Design and Scalable Assembly of High Density Low Tortuosity Electrodes. Massachusetts Institute of Technology. Cambridge, MA. Jun. 17, 2015. Believed to have been distributed on May 1, 2015. Presented on Jun. 17, 2015.

Cho et al., Self-Assembling Colloidal-Scale Devices: Selecting and Using Short-Range Surface Forces Between Conductive Solids. *Adv Funct Matter*. 2007;17:379-89.

Choi et al., Determination of electrochemical active area of porous Li_{1-x}CoO₂ electrode using the GITT technique. *Solid State Ionics*. 1998;109:159-63.

Chung et al., Validity of Bruggeman relation for porous electrodes. *Modelling and Simulation in Materials Science and Engineering*. 2013;21:1-16.

Darling et al., Modeling a Porous Intercalation Electrode with Two Characteristic Particle Sizes. *J Electrochem Soc.* 1997;144:4201-08.

Darling et al., Modeling Side Reactions in Composite Li₇Mn₂O₄ Electrodes. *J Electrochem Soc.* 1998;145:990-98.

Doyle et al., Analysis of capacity-rate data for lithium batteries using simplified models of the discharge process. *J Appl Electrochem*. 1997;27:846-56.

Doyle et al., Modeling of Galvanostatic Charge and Discharge of the Lithium/Polymer/Insertion Cell. *J Electrochem Soc.* 1993;140:1526-33.

Doyle, Design and Simulation of Lithium Rechargeable Batteries. Ph.D. Thesis. University of California, Berkeley. 1995. 370 pages.

Ehrlich, Chapter 35. *Lithium-Ion Batteries. Handbook of Batteries*. 3rd ed. 2002:35.1.

Elliott, S.R., "Physics of Amorphous Materials," Longman Group Limited. New York. 1984.

Erb et al., Composites reinforced in three dimensions by using low magnetic fields. *Science*. Jan. 13, 2012;335(6065):199-204. Doi: 10.1126/science.1210822.

Erb et al., Magnetic assembly of colloidal superstructures with multipole symmetry. *Nature*. Feb. 19, 2009;457(7232):999-1002. doi: 10.1038/nature07766.

Erb et al., Beyond diffusion-limited aggregation kinetics in microparticle suspensions. *Physical Review E*. Nov. 2, 2009; 80(5):051402.

Fermigier et al., Structure evolution in a paramagnetic latex suspension. *J Colloid Interface Science*. 1992;154(2):522-39.

Furst et al., Permanently Linked Monodisperse Paramagnetic Chains. *Langmuir*. 1998;14(26):7334-36.

French, R.H. et al., "Full Spectral Calculation of Non-Retarded Hamaker Constants for Ceramic Systems from Interband Transition Strengths," *Solid State Ionics* 75:13-33 (1995).

French, R.H. et al., "Origins and Applications of London Dispersion Forces and Hamaker Constants in Ceramics," *J. Am. Ceram. Soc.* 83(9):2117-46 (2000).

Fuller et al., Simulation and Optimization of the Dual Lithium Ion Insertion Cell. *J Electrochem Soc.* 1994;141:1-10.

Gaubicher et al., Li/β-VOPO₄: A New 4 V System for Lithium Batteries. *J Electrochem Soc.* 1999;146(12):4375-79.

Ghosh, S. et al., "Supramolecular Self-Assembly for Enhanced Conductivity in Conjugated Polymer Blends: Ionic Crosslinking in Blends of Poly(3,4-Ethylenedioxythiophene)-Poly(Styrenesulfonate) and Poly(vinylpyrrolidone)," *Adv. Mater.* 10(14):1097 (1998).

Gray, F.M., "Solid Polymer Electrolytes," VCH Publishers Inc. New York. (1991).

Guyomard et al., Li Metal-Free Rechargeable LiMn₂O₄/Carbon Cells: Their Understanding and Optimization. *J Electrochem Soc.* 1992;139:937-48.

(56)

References Cited

OTHER PUBLICATIONS

de l'orientation par diffraction des rayons X aux petits angles," *Colloid Poly Sci.*, 257:15-22, 1979.

Hart, B. W., et al., "3-D Microbatteries," *Electrochemistry Communications* 5:120-3 (2003).

Huang et al., Approaching Theoretical Capacity of LiFePO₄ at Room Temperature at High Rates. *Electrochem Solid-State Letts.* 2001;4(10):A170-72.

Idota, Y. et al., "Tin-Based Amorphous Oxide: A High-Capacity Lithium-Ion-Storage Material," *Science*, vol. 276 (May 30, 1997) pp. 1395-1397.

Kuwabata, S. et al., "Charge-Discharge Properties of Composites of LiMn₂O₄ and Polypyrrole as Positive Electrode Materials for 4 V Class of Rechargeable Li Batteries," *Electrochimica Acta* 44(23-26):4593-600 (1999).

Le Cras, F. et al., "Lithium intercalation in Li—Mg—Mn—O and Li—Al—Mn—O spinels," *Solid State Ionics*, 89 (1996) pp. 203-213.

Li et al., Li-Storage via Heterogeneous Reaction in Selected Binary Metal Fluorides and Oxides, *Journal of the Electrochemical Society*, Oct. 21, 2004. 151 (11): A1878-85. doi: 10.1149/1.1801451.

Li et al., Synthesis and properties of poly(2,5,8,11,14,17,20,23-octaoxapentacosyl methacrylate)-block-poly(4-vinylpyridine). *Makromol Chem.* 1991;192:3043-50.

Li et al., Project ES071: Design and Scalable Assembly of High Density Low Tortuosity Electrodes. Poster. Submitted on May 9, 2016. Presented on Jun. 9, 2016.

Limthongkul, P. et al., "Nanocomposite Li-Ion Battery Anodes Produced by the Partial Reduction of Mixed Oxides," *Chem Mat*, 13 (2001) pp. 2397-2402.

Linden, Editor-in-Chief, "Handbook of Batteries," Second Edition, McGraw-Hill, Inc., USA, 1995 (no. month), p. 2. 19.

Long, J.W., et al., "Three-Dimensional Battery Architectures," *Chem. Rev.* 104(10):4463-92 (2004).

Matsumoto, O., et al., "Ionic Conductivity of Dual-Phase Polymer Electrolytes Comprise of MBR-SBR Latex Films Swollen with Lithium Slat Solutions", *J. Electrochem Soc.*, vol. 141, No. 8, Aug. 1994, 1989-93.

Matsumoto, Osamu et al., "Vaporization of Graphite in Plasma Arc and Identification of C60 in the Deposit," *J. Electrochem. Soc.*, vol. 139, No. 1, Jan. 1992, L8-9.

Mazouzi et al., High capacity and excellent cyclability of Nano-Silicon/CB/CMC composite anode from tailored electrode preparation. 216th Electrochemical Society Meeting. 2009: Abstract #417.

Meredith et al., II. Conduction in Heterogeneous Systems. In: *Advances in Electrochemistry and Electrochemical Engineering*. John Wiley & Sons. New York. 1962;2:15-47.

Milling, A. et al., "Direct Measurement of Repulsive Van Der Waals Interactions Using an Atomic Force Microscope," *J. Colloid & Interface Science* 180:460-5 (1996).

Minett, M.G. et al., "Polymeric Insertion Electrodes," *Solid State Ionics*, 28-30 (1988), pp. 1192-1196.

Moon et al., Production of highly aligned porous alumina ceramics by extruding frozen alumina/camphene body. *Journal of European Ceramic Society*. 2011;31:1945-50.

Nagaoka, K. et al., "High Ionic Conductivity in Poly(dimethyl Siloxane-co-Ethylene Oxide) Dissolving Lithium Perchlorate," *J. Polymer Science*, pp. 659-663, vol. 22, 1984.

Nagarajan et al., A Mathematical Model for Intercalation Electrode Behavior. I. Effect of Particle-Size Distribution on Discharge Capacity. *J Electrochem Soc.* 1998;145:771-79.

Nakahara et al., Rechargeable batteries with organic radical cathodes. *Chemical Physical Letters*. Jun. 27, 2002. 359 (5-6): 351-4.

Neumann et al., "Negative Hamaker Coefficients," *Colloid and Polymer Science* 257(4):413-9 (1979).

Newman, Chapter 22. Porous Electrodes. In: *Electrochemical Systems*. 2nd Edition. Prentice Hall. Englewood Cliffs, New Jersey. 1991:450-95.

Nishide et al., Organic radical battery: nitroxide polymers as a cathode-active material. *Chem. Phys. Lett. Electrochim. Acta*. Nov. 30, 2004. 50(2-3): 827-31. doi:10.1016/j.electacta.2004.02.052.

Ohzuku, Tsutomu et al., "Synthesis and Characterization of LiAl_{1/3}ax;1;4Ni_{1/3}ax;3;4O₂ (R3m) for Lithium-Ion (Shuttlecock) Batteries," *J. Electrochem. Soc.*, vol. 142, No. 12, Dec. 1995, pp. 4033-4039.

Ong et al., Double-Layer Capacitance in Dual Lithium Ion Insertion Cell. *J Electrochem Soc.* 1999;146:4360-65.

Padhi et al., Phospho-olivines as Positive-Electrode Materials for Rechargeable Lithium Batteries. *J Electrochem Soc.* 1997;144(4):1188-94.

Pals et al., Thermal Modeling of the Lithium/Polymer Battery. I. Discharge Behavior of a Single Cell. *J Electrochem Soc.* 1995;142:3274-81.

Pals et al., Thermal Modeling of the Lithium/Polymer Battery. II. Temperature Profiles in a Cell Stack. *J Electrochem Soc.* 1995;142:3282-88.

Pierson, Chapter 6. Virtuous Carbon. *Handbook of Carbon, Graphite, Diamond and Fullerenes-Properties, Processing and Applications*. 1993;122-40.

Plitz et al., Structure and Electrochemistry of Carbon-Metal Fluoride Nanocomposites Fabricated by a Solid State Redox Conversion Reaction. *J. Electrochem. Soc.* 2005. 152(2): A307-15. doi: 10.1149/1.1842035.

Ravet et al., Electroactivity of natural and synthetic triphylite. *J Power Sci.* 2001;97-98:503-07.

Ravet et al., Improved Iron Based Cathode Material. Abstract No. 2014. 127.

Reed et al., Introduction to the Principles of Ceramic Processing. *J Wiley & Sons*, New York. 1988:86-89.

Sander, Jonathan Samuel, "Scalable method for channeled low tortuosity porosity in battery electrodes," Available online at: <http://p3.snf.ch/Project-148768>. Believed to have been made available on Jul. 7, 2013.

Sander, Jonathan Samuel, "Scalable method for channeled low tortuosity porosity in battery electrodes," Available online at: <http://p3.snf.ch/Project-154584>. Believed to have been made available on Jul. 1, 2014.

Sander et al., High-performance battery electrodes via magnetic templating. *Nature Energy*. Jul. 11, 2016;1: 16099.

Sander et al., Magnetic Transport, Mixing and Release of Cargo with Tailored Nanoliter Droplets. *Advanced Mater.* May 15, 2012;24(19):2582-7.

Soon et al., Assembling unidirectionally frozen alumina/camphene bodies for aligned porous alumina ceramics with larger dimensions. *Journal of European Ceramic Society*. 2011;31:415-9.

Tarascon et al., Issues and challenges facing rechargeable lithium batteries. *Nature*. Nov. 15, 2001;414(6861):359-67.

Thorat et al., Quantifying tortuosity in porous Li-ion battery materials. *Journal of Power Sources*. 2009;3(15):1-9.

Tukamoto et al., Electronic Conductivity of LiCoO₂ and its Enhancement by Magnesium Doping. *J Electrochem Soc.* 1997;144(9):3164-68.

Vaccaro et al., Electrochemical Investigations of Alkali-Metal Intercalation Reactions in TiS₂: Chronoamperometric Determination of Mass and Charge Transport Properties of Liquid Electrolyte Systems. *J Electrochem Soc.* 1982;129:682-88.

Van Oss, C.J. et al., "Applications of Net Repulsive Van Der Waals Forces Between Different Particles, Macromolecules or Biological Cells in Liquids," *Colloids and Surfaces* 1:45-56 (1980).

Van Oss, C.J. et al., "Comparison Between Antigen-Antibody Binding Energies and Interfacial Free Energies," *Immunological Communications* 6(4):341-54 (1977).

Van Oss, C.J. et al., "Repulsive Van Der Waals Forces. I. Complete Dissociation of Antigen-Antibody Complexes by Means of Nega-

Explore Litigation Insights

Docket Alarm provides insights to develop a more informed litigation strategy and the peace of mind of knowing you're on top of things.

Real-Time Litigation Alerts



Keep your litigation team up-to-date with **real-time alerts** and advanced team management tools built for the enterprise, all while greatly reducing PACER spend.

Our comprehensive service means we can handle Federal, State, and Administrative courts across the country.

Advanced Docket Research



With over 230 million records, Docket Alarm's cloud-native docket research platform finds what other services can't. Coverage includes Federal, State, plus PTAB, TTAB, ITC and NLRB decisions, all in one place.

Identify arguments that have been successful in the past with full text, pinpoint searching. Link to case law cited within any court document via Fastcase.

Analytics At Your Fingertips



Learn what happened the last time a particular judge, opposing counsel or company faced cases similar to yours.

Advanced out-of-the-box PTAB and TTAB analytics are always at your fingertips.

API

Docket Alarm offers a powerful API (application programming interface) to developers that want to integrate case filings into their apps.

LAW FIRMS

Build custom dashboards for your attorneys and clients with live data direct from the court.

Automate many repetitive legal tasks like conflict checks, document management, and marketing.

FINANCIAL INSTITUTIONS

Litigation and bankruptcy checks for companies and debtors.

E-DISCOVERY AND LEGAL VENDORS

Sync your system to PACER to automate legal marketing.